

Quantum Roaming in the complex forming mechanism of the reactions of OH with Formaldehyde and Methanol at low temperature and the effect of pressure

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The quantum dynamics of the title reactions is studied[1] using the ring polymer molecular dynamics (RPMD) method from 20 to 1200 K using full dimensional potential energy surfaces[2,3] which include long-range dipole– dipole interactions. A V-shaped dependence of the reaction rate constants is found with a minimum at 200–300 K, in rather good agreement with the current experimental data[4,5,6,7]. For temperatures above 300 K the reaction proceeds following a direct H-abstraction mechanism. However, below 100 K the reaction proceeds via organic-molecule... OH collision complexes, with very long lifetimes, longer than 10^{-7} s, associated with quantum roaming arising from the inclusion of quantum effects by the use of RPMD. New results are obtained in a new Neural-Network PES recently proposed, increasing the accuracy of the fit and including the long-range interactions[8]. The same trapping mechanism is found [8], and it is associated to spurious RPMD resonances[9]. Thermostated RPMD method[9] has been applied and compared to previous RPMD results. Also, preliminary results on the reaction of OH with the $(\text{H}_2\text{CO})_2$ dimer will be presented.

References

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